

4-(4-Ethylcyclohexyl)phenol

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Key indicators

Single-crystal X-ray study

$T = 294$ K

Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å

Disorder in main residue

R factor = 0.044

wR factor = 0.127

Data-to-parameter ratio = 15.3

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

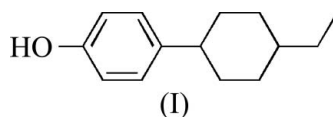
The crystal structure of the title compound, $\text{C}_{14}\text{H}_{20}\text{O}$, is stabilized by a disordered $\text{O}-\text{H}\cdots\text{O}$ hydrogen bond.

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Comment

Phenol derivatives are widely used as starting materials for many products, including drugs and liquid crystalline materials (Eidenschink *et al.*, 1978). The title compound, (I) (Fig. 1), has been synthesized from 4-(4-ethylcyclohexyl)acetophenone. The bond lengths and angles are within their normal ranges. The cyclohexyl ring adopts a chair conformation and the structure is stabilized by $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds (Table 1). The H atom of the hydroxy group was found to be disordered over two sites, with equal occupancies.



Experimental

The title compound was prepared according to the procedure of Hu *et al.* (2003). Suitable crystals of (I) (m.p. 400–402 K) were obtained by slow evaporation of a methanol solution.

Crystal data

$\text{C}_{14}\text{H}_{20}\text{O}$

$M_r = 204.30$

Monoclinic, $P2_1/c$

$a = 16.121$ (5) Å

$b = 5.4451$ (17) Å

$c = 13.478$ (4) Å

$\beta = 92.727$ (6)°

$V = 1181.7$ (6) Å³

$Z = 4$

$D_x = 1.148$ Mg m⁻³

Mo $K\alpha$ radiation

$\mu = 0.07$ mm⁻¹

$T = 294$ (2) K

Block, colourless

$0.24 \times 0.20 \times 0.18$ mm

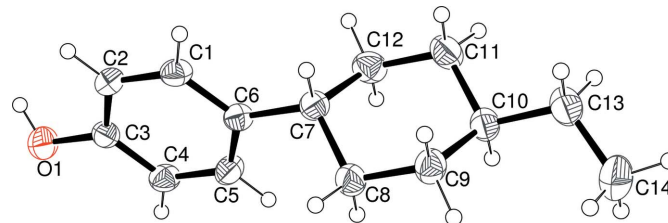


Figure 1

The molecular structure of (I), shown with 40% displacement ellipsoids. H atoms are shown as small spheres of arbitrary radii. Only one of the disordered H atoms attached to O1 is shown.

Data collection

Bruker SMART CCD diffractometer
 ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.978$, $T_{\max} = 0.988$

5722 measured reflections
 2101 independent reflections
 1198 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.040$
 $\theta_{\text{max}} = 25.0^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.044$
 $wR(F^2) = 0.127$
 $S = 1.01$
 2101 reflections
 137 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0609P)^2 + 0.0454P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.14 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.14 \text{ e } \text{\AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O1-H1A\cdots O1^i$	0.84	2.05	2.893 (3)	180
$O1-H1B\cdots O1^{ii}$	0.84	2.11	2.953 (3)	179

Symmetry codes: (i) $-x, -y, -z$; (ii) $-x, -y + 1, -z$.

To avoid an unreasonable intermolecular $H\cdots H$ contact, the H atom of the OH group must be disordered over two positions with equal occupancy so as to form alternative $O-H\cdots O$ bonds to adjacent OH groups. These atoms were positioned geometrically and refined as riding. All C-bound H atoms were positioned geometrically ($C-H = 0.93-0.97 \text{ \AA}$) and refined using a riding model, with $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$ or $1.5U_{\text{eq}}(\text{methyl C})$.

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

References

Bruker (1997). SMART, SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
 Eidenschink, R., Krause, J. & Pohl, L. (1978). US Patent No. 4 130 502.
 Hu, B. H., Xia, Y. T., Zhou, Y. B., Meng, F. M., Chen, X. & Fu, W. G. (2003). Chinese Patent No. 1 463 961.
 Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
 Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.